

Coûtage type

D'examen de contrôle de 3<sup>ème</sup> année  
chimie fondamentale

Module : chimie thermodynamique des  
Solutions.

08,00

exercice N°1

$$\begin{aligned}
 PV &= nRT \Rightarrow \frac{V}{n} = \frac{RT}{P} \\
 &= V_m = \frac{RT}{P} \\
 &= V_m = \frac{0,08206 \cdot 298}{0,987} \\
 &= 24,77 \text{ l/mol}
 \end{aligned}$$

$$\begin{aligned}
 2 - \rightarrow U &= Q + W \Rightarrow dU = dQ + dW \\
 &\Rightarrow dU = Tds - PdV \\
 &\Rightarrow \frac{dU}{n} = T \frac{ds}{n} - P \frac{dV}{n} \\
 &\Rightarrow dU_m = T ds_m - P dV_m
 \end{aligned}$$

$$\begin{aligned}
 * H &= U + PV \Rightarrow dH = dU + PdV + VdP \\
 &\Rightarrow dH = Tds - PdV + PdV + VdP \\
 &\Rightarrow dH = Tds + VdP \\
 &\Rightarrow \frac{dH}{n} = T \frac{ds}{n} + \frac{V}{n} dP \\
 &\Rightarrow dH_m = T ds_m + V_m dP
 \end{aligned}$$

$$\begin{aligned}
 * F &= U - TS \Rightarrow dF = dU - Tds - SdT \\
 &\Rightarrow dF = Tds - PdV - Tds - SdT \\
 &\Rightarrow dF = -PdV - SdT \\
 &\Rightarrow \frac{dF}{n} = -P \frac{dV}{n} - \frac{S}{n} dT \\
 &\Rightarrow dF_m = -P dV_m - S_m dT
 \end{aligned}$$

$$\begin{aligned}
 * G &= H - TS \Rightarrow dG = dH - Tds - SdT \\
 &\Rightarrow dG = Tds + VdP - Tds - SdT \\
 &\Rightarrow dG = VdP - SdT \\
 &\Rightarrow \frac{dG}{n} = \frac{V}{n} dP - \frac{S}{n} dT \\
 &\Rightarrow \frac{dG}{n} = \frac{V}{n} dP - \frac{S}{n} dT
 \end{aligned}$$

$$3 - \left[ \frac{\partial}{\partial T} \left( \frac{\mu_i}{T} \right) \right]_{P, n_i} = \frac{\partial \mu_i}{\partial T} \cdot \frac{1}{T} - \frac{\partial T}{\partial T} \frac{\mu_i}{T^2}$$

$$\begin{aligned}
 dG_i &= d\mu_i = V dP - S dT \\
 d\mu_i &= -S dT
 \end{aligned}$$

$$\text{donc } \left[ \frac{\partial}{\partial T} \left( \frac{\mu_i}{T} \right) \right]_{P, n_i} = \frac{-S T - \mu_i}{T^2}$$

$$\begin{aligned}
 \mu_i &= G_i = H_i - TS \\
 \frac{\partial}{\partial T} \left( \frac{\mu_i}{T} \right) &= \frac{-S T - H_i + TS}{T^2}
 \end{aligned}$$

$$\left[ \frac{\partial}{\partial T} \left( \frac{\mu_i}{T} \right) \right]_{P, n_i} = \frac{-H_i}{T^2}$$

suite de question 2.  
la relation de G, F, U, H  
avec le potentiel chimique  
 $dG = VdP - SdT + \mu^* dn$   
 $dF = -PdV - SdT + \mu^* dn$   
 $dH = Tds + VdP + \mu^* dn$   
 $dU = Tds - PdV + \mu^* dn$